SAr

Air Oxidative Radical Oxysulfurization of Alkynes Leading to α -Thioaldehydes

Shao-Fang Zhou,† Xiang-Qiang Pan,*,† Zhi-Hao Zhou,† Adedamola Shoberu,† Pei-Zhi Zhang,† and Jian-Ping Zou*,†,‡

† Key Laboratory of Or[gan](#page-5-0)ic Synthesis of Jiangsu Province, College of Chemistry and Chemical Engineering, Soochow University, 199 Renai Street, Suzhou, Jiangsu 215123, China

 $ArSH + R$

‡ Key Laboratory of Synthetic Chemistry of Natural Substances, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai, Shanghai 200032, China

S Supporting Information

[AB](#page-5-0)STRACT: [Air oxidative](#page-5-0) radical oxysulfurization of alkynes initiated by 0.5 mol % tert-butyl hydroperoxide with arylthiols is described. The reaction proceeded at room temperature in the presence of 5% mol water to afford selective α -thioaldehydes.

> 1, entries 11−20), the optimum reaction conditions were determined to be thiophenol (1a, 2.0 equiv) and tert-butyl 2- [\(b](#page-1-0)ut-3-yn-1-yloxy) acetate $(2a, 1$ equiv) in THF at 25 °C for 48 h in the presence of 0.5% mol TBHP and 5% mol H_2O to afford the selective α -thioaldehyde 4a in good yield (Table 1, entry 12).

0.5 mol% TBHP

THF, 25°C, air, 48 h R 16 examples up to 91% yield

5 mol% $H₂O$

Under these conditions, reactions of a variety of terminal alkynes 2a−i with thiophenol (1a) were carried o[ut](#page-1-0). The alkynes tert-butyl 2-(but-3-yn-1-yloxy) acetate $(2a)$ and *n*-hexyne $(2b)$ gave exclusively α -thioaldehydes (4a and 4b) in excellent yields (Table 2, entries 1 and 2). However, the reaction of other chain terminal alkynes 2c−2g gave mainly α-thioaldehydes 4c−4g in moder[ate](#page-2-0) to good yields accompanied by alkenylsufides 3c−3g (Table 2, entries 3−7). In reactions involving the sterically hindered terminal alkynes such as *t*-butylacetylene $(2h)$ and cycl[op](#page-2-0)ropyl acetylene (2i), low yields of α -thioaldehydes 4h–4i were observed as the formation of alkenylsufides 3h−3i tends to predominate (Table 2, entries 8 and 9). The low yields of α thioaldehydes underline the effect of bulky group attached to the $C\equiv C$ triple bond. [It](#page-2-0) is worth noting that the reaction of phenylacetylene (2j) with 1a led to formation of complicated mixtures (Table 2, entry 10). This is probably due to the high reactivity of the conjugated alkenyl radical generated from the reaction of thiyl [ra](#page-2-0)dical with 2j. Also, when alkyl group $R = n$ butyl (2b) was replaced with phenylethyl (2k) and phenylpropyl (2l), their reactions with 1a produced 4k and 4l in low yields, 12% and 44%, respectively (Table 2, entries 11 and 12), and it was observed that most of starting substrate 2k and 2l remained unreacted at the end of reaction. F[ro](#page-2-0)m the results obtained, we reasoned that the yield of α -thioaldehydes depended mainly on the structure of the terminal alkynes.

Afterward, the reactions of terminal alkynes having oxygen or nitrogen-containing substituent groups were investigated. Alkynes bearing an ester group at α - and β -positions all reacted

Recently, difunctionalization of alkenes and alkynes has
become a powerful tool in synthetic organic chemistry.¹ The radical difunctionalization having advantages such as mild reaction conditions, high selectivity, and convenient workup h[as](#page-5-0) made it more interesting.² Thiol−oxygen co-oxidation reactions (TOCO) provide attractive routes to functionalized valuable produ[c](#page-5-0)ts with studies focusing mainly on olefins; 3 only a few have been on alkynes.^{3a,4} The first example of TOCO of alkynes appea[r](#page-6-0)ed in the early 1960s when Griesbaum et al.^{4b} reported the reaction of thiophe[nol](#page-6-0) with phenylacetylene in an oxygen atmosphere to give the phenylglyoxal hemithio[ace](#page-6-0)tal. In 1993, the Isoe group found that the electroinitiated oxygenation of alkynes in the presence of thiophenol gave α -(phenylthio) carbonyl compounds. $4c$ Also, Renaud et al. reported that the reaction of terminal alkynes with thiophenol in the presence of AIBN afforded a serie[s o](#page-6-0)f five-membered cyclic compounds.^{4e−g} In continuation of our efforts on difunctionalization reactions,⁵ herein, we report a new protocol, TBHP (tert-butyl hy[dro](#page-6-0)peroxide)-initiated water-catalyzed difunctionalization of term[i](#page-6-0)nal alkynes with thiophenol in air at room temperature to give α thioaldehydes.

We began our studies by exploring the reaction of thiophenol (1a) with tert-butyl 2-(but-3-yn-1-yloxy) acetate (2a) in DMF at 25 °C; however, alkenyl sulfide 3a was obtained as the sole product (Table 1, entry 1). After screening of solvents, we were pleased to observe a mixture of products 3a and 4a in tert-butanol and ethereal s[ol](#page-1-0)vents such as 1,4-dioxane, tetrahydrofuran (THF), and 1,2-dimethoxyethane (1,2-DME) (Table 1, entries 5−8). Interestingly, the addition of a catalytic amount of water into the reaction in THF slightly increased the yield of [4a](#page-1-0) (Table 1, entry 9). More interesting was the fact that no formation of 4a was observed in anhydrous THF (Table 1, entry 10). To improve [th](#page-1-0)e selectivity and yield of reaction, 0.5% mol of TBHP was added to the reaction and this led to a [m](#page-1-0)assive increase in the yield of 4a to 75% (Table 1, entry 11). After screening the reaction time, temperature, amount of water. and TBHP (Table

Received: March 22, 2015 Published: April 29, 2015

Table 1. Optimization of the Reaction Conditions

 a Isolated yield. b Addition of 5% mol H₂O (1 μ L). c Anhydrous THF. d Gas chromatography yield. e Addition of 0.5% mol TBHP (1 μ L). f Addition of 4 μL H₂O. ^gAddition of 5% mol TBHP (10 μL).

well with 1a to give exclusively alkenylsufides 3m−3n (Table 2, entries 13 and 14). Reactions involving 5-hydroxypentyne (2o) and 4-hydroxybutyne (2p) also gave alkenylsufides 3o and 3p [as](#page-2-0) the sole products in 77% and 17% yield, respectively (Table 2, entries 15 and 16). Interestingly, the presence of the oxygen atom and COO'Bu groups in [te](#page-2-0)rt-butyl 2-(but-3-yn-1-yloxy) acetate (2a) led to formation of α -thioaldehyde, although the reason for this remained unknown. Furthermore, no reactions were observed with α -aminoalkynes (2q−2s) (Table 2). We reasoned that the amino group on the α -carbon could probably be impeding the radical reaction.

To know the effect of substitution on pheny[l](#page-2-0) [r](#page-2-0)ing, a series of arylthiols containing electron-donating and electron-withdrawing groups were employed. In the reaction of 4-methylbenzenethiol (1b) with terminal alkyne 2a, α -thioaldehyde (4ba) was obtained as the major product in 59% yield and alkenylsulfide (3ba) as minor product in 32% yield (Table 3, entry 1). On the contrary, the reaction of 2-methylbenzenethiol (1c) gave the alkenylsulfide (3ca) as major product in [70](#page-3-0)% yield and α thioaldehyde (4ca) as minor product in 23% yield (Table 3, entry 2). In addition, the reaction of 2,6-dimethylbenzenethiol (1d) with 2a gave alkenylsulfide (3da) as sole product in 84[%](#page-3-0) yield (Table 3, entry 3). These results show that hindered arylthiyl radicals favor formation of the less hindered alkenylsulfides 3. 4- Methox[yb](#page-3-0)enzenethiol (1e) and 2-aminobenzenethiol (1f) did not give desired compounds due to self-coupling of the electronrich arylthiyl radicals generated from 1e and 1f leading to diaryldisulfides (Table 3, entries 4 and 5). Arylthiols bearing F and Cl groups reacted with 2a to form α -thioaldehydes (4ga, 4ha, 4ia and 4ja) as maj[or](#page-3-0) product in moderate to excellent yields (Table 3, entries 6−9). No reactions were observed with 4nitrobenzenethiol 1k, pyridylthiol 1l and furylthiol 1m (Table 3, entries 10−12). This is because the corresponding thiyl radicals could not be formed under the reaction condition.

As mentioned earlier, the reaction of tert-butyl 2-(but-3-yn-[1](#page-3-0) yloxy) acetate $(2a)$ with thiophenol $(1a)$ did not take place in anhydrous THF. Further experiment also showed that the same reaction did not take place when conducted in a nitrogen atmosphere, thus indicating that oxygen is necessary for reaction to occur. To understand how the reaction proceeds, a mechanistic study was done. An experiment using deuteriumlabeled thiophenol $(1a')$ with *n*-hexyne $(2b)$ was carried out and the unlabeled α -thioaldehyde (4b) was isolated as the product (Scheme 1, eq 1). However, the reaction of thiophenol (1a) with deuterium-labeled *n*-hexyne (2b[']) gave the deuterated α thioaldeh[yd](#page-3-0)e $(4b')$ (Scheme 1, eq 2), thus indicating that the aldehydic proton originated from *n*-hexyne $(2b)$. To know if the reaction proceeds via a radical [p](#page-3-0)athway, the reaction of tert-butyl 2-(but-3-yn-1-yloxy) acetate $(2a)$ with thiophenol $(1a)$ was carried out in the presence of the radical inhibitor, TEMPO (2,2,6,6-tetramethyl-1-piperidinyloxy) (5), and only 2,2,6,6 tetramethyl-1-((phenylthio)oxy)piperidine (6) and diphenyl disulfide (7) were isolated, with no detectable sign of α thioaldehyde (4a). Further experiment showed that TEMPO did not react with 2a.

On the basis of above findings, a plausible mechanism is proposed in Scheme 2. The thiophenyl radical 8 initiated by TBHP coordinates with terminal alkyne 2a to form flexible radical 9, followed by [ad](#page-4-0)dition of H_2O to give radical 10 which then reacts with O_2 (air) to give peroxy radical 11; this decomposes to form α -thioaldehyde 4a and hydroperoxy radical

a
Reaction conditions: thiophenol (1a, 2.0 equiv), tert-butyl 2-(but-3-yn-1-yloxy) acetate (2a, 1 equiv) in THF at 25 °C for 48 h in the presence of 0.5% mol TBHP and 5% mol H2O. ^b Isolated yield. ^c Analyzed by gas chromatography. ^d Mixture of alkenylsufide and diphenyldisulfide, they cannot be separated from each other. "Most of starting material 2k did not react with 1a.^fSome of starting material 2l did not react with 1a. ⁸N.R. represents no reaction.

12, which reacts with thiophenol (1a) to liberate thiophenyl radical 8, for further propagation of the reaction cycle.

In conclusion, a new protocol for the oxysulfurization of alkynes via the reaction of arylthiols with alkynes has been developed. The reaction was initiated by 0.5 mol % TBHP at

room temperature in the presence of 5 mol % water, with air (O_2) as sole oxidant to afford the selective α -thioaldehydes in moderate to good yields. This method is straightforward, requires no other oxidant or additive, and involves simple manipulations. The α -thioaldehydes obtained can be directly

a
Reaction conditions: arylthiol (1, 2.0 equiv), tert-butyl 2-(but-3-yn-1-yloxy) acetate (2a, 1 equiv) in THF at 25 °C for 48 h in the presence of 0.5% mol TBHP and 5% mol H₂O. ^bIsolated yield. ^cN.R. indicates that the alkyne is not consumed and the desired α -thioaldehyde and/or alkenylsulfide mol TBHP and 5% mol H₂O. ^bIsolated yield. ^cN.R. indicates that were not isolated (0% yield).

Scheme 1. Mechanistic Study

applied in syntheses of organic, medicinal and other functional compounds.

EXPERIMENTAL SECTION

General Methods. ¹H NMR (400 MHz) and ¹³C NMR (101 MHz) spectra were determined with CDCl₃ or DMSO- d_6 as solvent and tetramethylsilane (TMS) as internal standard. Chemical shifts were reported in parts per million (ppm) from internal TMS (δ) ; all coupling constants (J values) were reported in hertz (Hz). High-resolution mass spectra were recorded on a TOF machine (ESI). Column chromatography was performed with 300−400 mesh silica gel using flash column techniques. All of the reagents were used directly as obtained commercially unless otherwise noted.

Preparation of 2-(Arylthio) Aldehydes 4. Typical Procedure for the Preparation of tert-Butyl 2-(4-Oxo-3-(phenylthio)butoxy) Acetate (4a). To a solution of THF (10 mL), tert-butyl 2-(but-3-yn-1 yloxy) acetate (2a, 0.184 g, 1 mmol) and thiophenol (0.22 g, 2 mmol) was added TBHP (1 μ L) and H₂O (1 μ L), and the mixture was stirred at 25 \degree C for 48 h. After the completion of the reaction, the solvent was evaporated under vacuum to yield the crude product, which was purified by column chromatography (silica gel, petroleum ether/EtOAc = $40:1$) to give tert-butyl 2-(4-oxo-3-(phenylthio)butoxy) acetate (4a).

tert-Butyl 2-(4-oxo-3-(phenylthio)butoxy) Acetate (4a). Yellow oil, 72% yield (223 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.52 (d, J = 2.8 Hz, 1H), 7.42−7.40 (m, 2H), 7.34−7.23 (m, 3H), 3.93 (d, J = 3.3 Hz, 2H), 3.90−3.82 (m, 1H), 3.78−3.62 (m, 2H), 2.24−2.12 (m, 1H), 1.94−1.86 (m, 1H), 1.47 (s, 9H). ¹³C NMR (101 MHz, CDCl₃): δ 189.9, 164.9, 129.0, 126.6, 124.6, 123.9, 77.2, 64.2, 63.5, 49.2, 23.7, 23.5. HRMS (ESI) m/z : $(M + Na)^+$ Calcd for $C_{16}H_{22}O_4$ SNa 333.1137, found 333.1134.

2-(Phenylthio)hexanal (4b). e^{zz} Yellow oil, 78% yield (162 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.34 (d, J = 4.3 Hz, 1H), 7.38–7.34 (m, 2H), 7.30−7.24 (m, 3H), 3.52[−](#page-6-0)3.46 (m, 1H), 1.84−1.74 (m, 1H), 1.71−1.59 (m, 1H), 1.56−1.28 (m, 4H), 0.90 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, DMSO-d₆): δ 190.8, 128.2, 127.3, 124.6, 123.6, 52.3, 24.5, 23.0, 17.8, 9.4, 9.2. MS (ESI) m/z : (M + H)⁺ Calcd for C₁₂H₁₇OS 209.1, found 209.1.

2-(Phenylthio)pentanal (4c).⁷ Yellow oil, 64% yield (124 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.37 (d, J = 4.3 Hz, 1H), 7.42–7.35 (m, 2H), 7.33−7.26 (m, 3H), 3.58[−](#page-6-0)3.50 (m, 1H), 1.80−1.74 (m, 1H), 1.71−1.42 (m, 3H), 0.97 (t, J = 7.3 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃): δ 195.3, 132.8, 131.9, 129.2, 128.1, 56.6, 29.9, 20.2, 13.8. MS (ESI) m/z : $(M + H)^+$ Calcd for $C_{11}H_{15}OS$ 195.1, found 195.1.

2-(Phenylthio)heptanal (4d).⁸ Yellow oil, 71% yield (157 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.36 (d, J = 2.1 Hz, 1H), 7.40–7.36 (s, 2H), 7.30−7.22 (m, 3H), 3.52−3.50 ([m](#page-6-0), 1H), 1.91−1.73 (m, 1H), 1.73−1.61 (m, 1H), 1.61−1.52 (m, 1H), 1.49−1.39 (m, 1H), 1.20−1.15 (m, 4H), 0.89 (s, 3H). 13 C NMR (75 MHz, CDCl₃): δ 195.3, 132.7, 131.9, 129.1, 128.1, 56.8, 31.4, 27.8, 26.6, 22.4, 14.0. MS (ESI) m/z: (M + H)+ Calcd for $C_{13}H_{19}OS$ 223.1, found 223.1.

2-(Phenylthio) octanal (4e). Yellow oil, 68% yield (160 mg) ; ¹H NMR (400 MHz, CDCl₃): δ 9.36 (d, J = 4.3 Hz, 1H), 7.38 (dd, J = 7.5, 2.0 Hz, 2H), 7.33−7.27 (m, 3H), 3.52 (td, J = 7.4, 4.3 Hz, 1H), 1.87− 1.75 (m, 1H), 1.73−1.59 (m, 1H), 1.52−1.38 (m, 2H), 1.36−1.28 (m, 6H), 0.88 (t, J = 6.8 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃): δ 195.3, 132.7, 131.9, 129.1, 128.1, 56.8, 31.5, 28.9, 27.8, 26.9, 22.5, 14.0. HRMS (ESI) $m/z\mathrm{:}~(M+H)^+$ Calcd for $\mathrm{C}_{14}\mathrm{H}_{21}\mathrm{OS}$ 237.1313, found 237.1323.

2-(Phenylthio)nonanal (4f). Yellow oil, 64% yield (160 mg) ; ¹H NMR (400 MHz, CDCl₃): δ 9.36 (d, J = 4.1 Hz, 1H), 7.40–7.36 (m, 2H), 7.32−7.28 (m, 3H), 3.54−3.48 (m, 1H), 1.82−1.76 (m, 1H), 1.72−1.60 (m, 1H), 1.60−1.38 (m, 2H), 1.32−1.28 (m, 8H), 0.87 (t, J = 6.6 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃): δ 195.3, 132.7, 131.9, 129.1, 128.1, 56.8, 31.7, 29.2, 29.0, 27.8, 26.9, 22.6, 14.1. HRMS (ESI) m/z: (M $+ H$ ⁺ Calcd for C₁₅H₂₃OS 251.1470, found 251.1515.

2-(Phenylthio)decanal (4g). Yellow oil, 60% yield (158 mg) ; ¹H NMR (400 MHz, CDCl₃): δ 9.34 (d, J = 4.3 Hz, 1H), 7.41–7.32 (m, 2H), 7.28−1.24 (m, 3H), 3.49 (td, J = 7.3, 4.4 Hz, 1H), 1.84−1.74 (m, 1H), 1.68−1.60 (m, 1H), 1.56−1.36 (m, 2H), 1.26−1.20 (m, 10H), 0.86 (t, J = 6.7 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃): δ 195.3, 132.7, 131.9

 $(d, J = 7.1 \text{ Hz}, 2H), 7.27 \text{ (dd, } J = 15.4, 7.5 \text{ Hz}, 3H), 3.24 \text{ (d, } J = 6.4 \text{ Hz},$ 1H), 1.17 (s, 9H). 13C NM, 129.1, 128.1, 56.8, 31.8, 29.3, 29.2, 29.1, 27.8, 26.9, 22.6, 14.1. HRMS (ESI) m/z : $(M + H)^+$ Calcd for $C_{16}H_{25}OS$ 265.1626, found 265.1622.

3,3-Dimethyl-2-(phenylthio)butanal (4h). Yellow oil, 51% yield (106 mg) ; ¹H NMR (400 MHz, CDCl₃): δ 9.47 (d, J = 6.4 Hz, 1H), 7.37 R (101 MHz, CDCl₃): δ 195.9, 132.8, 129.9, 128.4, 69.3, 34.3, 28.5. HRMS (ESI) m/z : $(M + K)^+$ Calcd for $C_{12}H_{16}OSK$ 247.0559, found 247.0542.

2-Cyclopropyl-2-(phenylthio)acetaldehyde (4i). Yellow oil, 52% yield (100 mg) ; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 9.42 $(\text{d}, \text{J} = 4.5 \text{ Hz}, 1 \text{ H})$, 7.40−7.34 (m, 2H), 7.30−7.24 (m, 3H), 2.91−2.83 (m, 1H), 1.05−0.95 (m, 1H), 0.78−0.66 (m, 2H), 0.53−0.38 (m, 2H). 13C NMR (75 MHz, CDCl₃): δ 194.5, 133.1, 129.1, 129.0, 128.1, 127.5, 127.1, 62.1, 9.4, 5.7, 4.3. HRMS (ESI) m/z : $(M + H)^+$ Calcd for $C_{11}H_{13}$ OS 193.0687, found 193.0681.

4-Phenyl-2-(phenylthio)butanal (4k). Yellow oil, 12% yield (30 mg); ¹H NMR (400 MHz, CDCl3): δ 9.52 (d, J = 3.4 Hz, 1H), 7.46 (ddd, J = 6.5, 3.8, 1.5 Hz, 2H), 7.41−7.35 (m, 5H), 7.32−7.25 (m, 3H), 3.56 (td, J = 13.7, 6.9 Hz, 1H), 3.00−2.82 (m, 2H), 2.30−2.15 (m, 1H), 2.12−1.96 (m, 1H). ¹³C NMR (101 MHz, CDCl₃): δ 194.4, 139.9, 132.8, 130.8, 128.7, 128.16, 128.10, 127.9, 125.9, 55.5, 32.2, 28.7. HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₆H₁₆OS 256.0922, found 256.0914.

5-Phenyl-2-(phenylthio)pentanal (4l). Yellow oil, 44% yield (120 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.37 (d, J = 4.0 Hz, 1H), 7.37− 7.32 (m, 2H), 7.32−7.26 (m, 5H), 7.22−7.16 (m, 3H), 3.51 (td, J = 6.9, 4.0 Hz, 1H), 2.70−2.64 (m, 2H), 1.96−1.58 (m, 4H). 13C NMR (75 MHz, CDCl₃): δ 195.3, 141.6, 136.7, 133.3, 131.8, 130.1, 129.4, 129.3, 128.69, 128.66, 128.5, 127.8, 127.4, 126.3, 57.1, 35.7, 28.8, 27.5. HRMS (ESI) m/z : $(M + Na)^+$ Calcd for C₁₇H₁₈OSNa 293.0976, found 293.0967.

Ethyl (Z/E)-3-(Phenylthio)acrylate (**3m**). 9 Yield 77% (150 mg); $^1\mathrm{H}$ NMR (400 MHz, CDCl₃): δ Z-isomer 7.78 (d, J = 14.9 Hz, 1H), 7.53– 7.45 (m, 2H), 7.42−7.36 (m, 3H), 5.91 (d, [J](#page-6-0) = 10.2 Hz, 1H), 4.26 (q, 2H), 1.33 (t, J = 7.2 Hz, 3H); E-isomer 7.78 (d, J = 14.9 Hz, 1H), 7.53– 7.45 (m, 2H, ArH), 7.42–7.36 (m, 3H), 5.65 (d, J = 14.9 Hz, 1H), 4.16 $(q, 2H)$, 1.26 $(t, J = 7.0 \text{ Hz}, 3H)$. MS (ESI) m/z : $(M + H)^+$ Calcd for $C_{11}H_{13}O_2S$ 209.1, found 209.1.

 (Z/\widetilde{E}) -3-(Phenylthio)allyl Acetate (3n). 10 Yield 70% (146 mg); 1 H NMR (400 MHz, CDCl₃): δ Z isomer 7.47–7.30 (m, 5H), 6.52 (dt, J = 15.1, 1.2 Hz, 1H), 6.00−5.73 (m, 1H), 4.[83](#page-6-0) (dd, J = 6.5, 1.2 Hz, 2H), 2.15 (s, 3H); E isomer 7.47–7.30 (m, 5H), 6.58 (dt, J = 15.1, 1.2 Hz, 1H), 6.00−5.73 (m, 1H), 4.65 (dd, J = 6.6, 1.2 Hz, 2H), 2.12 (s, 3H). HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₁H₁₂O₂S 208.0558, found 208.0558.

(Z/E) 5-(Phenylthio)pent-4-en-1-ol (3o). Yield 77% (149 mg) ; 1 H NMR (400 MHz, CDCl₃): δ Z isomer 7.42–7.31 (m, 4H), 7.28–7.21 $(m, 1H)$, 6.29 (dt, J = 9.2, 1.2 Hz, 1H), 5.88 (dt, J = 9.2, 7.4 Hz, 1H), 3.73 $(id, J = 6.4, 4.1 Hz, 2H), 2.40 (qd, J = 7.4, 1.2 Hz, 2H), 1.86 (s, 1H),$ 1.81−1.70 (m, 2H); E isomer 7.42−7.31 (m, 4H), 7.28−7.21 (m, 1H), 6.24 (dt, J = 14.9, 1.3 Hz, 1H (E)), 6.12–5.98 (m, 1H (Z)), 3.73 (td, J = 6.4, 4.1 Hz, 2H), 2.35−2.27 (m, 2H), 1.86 (s, 1H), 1.81−1.70 (m, 2H). 13C NMR (101 MHz, CDCl3): ^Z ⁺ ^E ^δ 136.0, 132.3, 129.03, 128.98, 128.91, 128.6, 126.3, 126.2, 123.7, 121.7, 62.2, 62.1, 31.9, 31.8, 29.4, 25.4. HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₁H₁₄OS 194.0765, found 194.0763.

(Z/E) 4-(Phenylthio)but-3-en-1-ol $(3p)^{11}$ Yellow oil, 17% yield (30) mg); ¹H NMR (400 MHz, CDCl₃): δ Z isomer 7.43–7.30 (m, 4H), 7.27−7.23 (m, 1[H\)](#page-6-0), 6.41 (d, J = 9.3 Hz, 1H), 6.03−5.84 (m, 1H), 3.75 $(t, J = 6.3 \text{ Hz}, 2H)$, 2.59 (td, J = 7.4, 1.0 Hz, 2H), 1.96 (s, 1H); E isomer 7.43−7.30 (m, 4H), 7.27−7.23 (m, 1H), 6.33 (d, J = 15.0 Hz, 1H), 6.03–5.84 (m, 1H), 3.79 (t, J = 6.4 Hz, 2H), 2.53–2.42(m, 2H), 1.96 (s, 1H). HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₀H₁₂OS 180.0609, found 180.0599.

(Z/E) tert-Butyl 2-((4-(p-tolylthio)but-3-en-1-yl)oxy)acetate (3ba). Yellow oil, 70% yield (215 mg); ¹H NMR (300 MHz, CDCl₃): δ 7.25 (d, $J = 8.0$ Hz, 2H), 7.11 (d, $J = 7.6$ Hz, 2H), 6.24 (t, $J = 11.4$ Hz, 1H), 6.02– 5.71 (m, 1H), 4.08−3.88 (m, 2H), 3.72−3.46 (m, 2H), 2.56 (dd, J = 13.4, 6.7 Hz, 1H), 2.46 (dd, J = 13.4, 6.6 Hz, 1H), 2.32 (s, 3H), 1.48 (s, 9H). ¹³C NMR (75 MHz, CDCl₃): Z + E δ 168.7, 168.6, 135.5, 135.4,

131.3, 131.0, 129.4, 128.7, 128.6, 128.5, 126.7, 125.0, 123.6, 80.6, 80.5, 69.6, 69.3, 67.8, 67.7, 32.3, 28.5, 27.1, 20.0. HRMS (CI-TOF) m/z : (M + H)⁺ Calcd for C₁₇H₂₅O₃S 309.1524, found 309.1526.

tert-Butyl 2-(4-oxo-3-(p-tolylthio)butoxy)acetate (4ba). Yellow oil, 23% yield (75 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.51 (d, J = 2.0 Hz, 1H), 7.29 (d, J = 7.8 Hz, 2H), 7.10 (d, J = 7.8 Hz, 2H), 3.98−3.88 (m, 2H), 3.78 (td, J = 7.3, 2.2 Hz, 1H), 3.74−3.62 (m, 2H), 2.32 (s, 3H), 2.18−2.10 (m, 1H), 1.92−1.82 (m, 1H), 1.48 (s, 9H). 13C NMR (75 MHz, CDCl₃): δ 193.4, 168.5, 137.9, 133.3, 128.9, 128.8, 127.5, 126.0, 107.7, 80.7, 67.8, 67.1, 53.0, 27.1, 20.1. HRMS (CI-TOF) m/z: M⁺ Calcd for $C_{17}H_{24}O_4S$ 324.1395, found 324.1395.

(Z/E) tert-Butyl 2-((4-(o-tolylthio)but-3-en-1-yl)oxy)acetate (3ca). Yellow oil, 32% yield (99 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.32− 7.30 (m, 1H), 7.18−7.10 (m, 3H), 6.24−6.14 (m, 1H), 5.99−5.82 (m, 1H), 4.0−3.96 (m, 2H), 3.68−3.56 (m, 2H), 2.65−2.55 (m, 1H), 2.54− 2.45 (m, 1H), 2.38 (s, 1.5H, (Z isomer)), 2.35 (s, 1.5 H, (E isomer)), 1.48 (s, 9H). ¹³C NMR (101 MHz, CDCl₃): δ Z + E 169.1, 137.09, 137.06, 134.7, 134.3, 130.9, 129.7, 128.83, 128.77, 128.6, 126.06, 126.05, 126.03, 125.9, 124.4, 122.9, 81.13, 81.09, 70.2, 69.9, 68.3, 68.2, 32.9, 29.0, 27.6, 20.0, 19.7. HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₇H₂₄O₃S 308.1446, found 308.1431.

tert-Butyl 2-(4-oxo-3-(o-tolylthio)butoxy)acetate (4ca). Yellow oil, 59% yield (191 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.43 (d, J = 3.4 Hz, 1H), 7.43−7.36 (m, 1H), 7.24−7.09 (m, 3H), 3.99−3.87 (m, 2H), 3.88−3.80 (m, 1H), 3.70 (td, J = 5.2, 1.3 Hz, 2H), 2.45 (s, 3H), 2.33− 2.14 (m, 1H), 2.07−1.83 (m, 1H), 1.47 (s, 9H). 13C NMR (101 MHz, CDCl₃): δ 194.5, 169.4, 140.4, 133.4, 131.2, 130.7, 128.3, 126.7, 81.8, 68.8, 68.2, 53.5, 28.8, 28.1, 20.9. HRMS (CI-TOF) m/z: (M + H)⁺ Calcd for $C_{17}H_{24}O_4S$ 324.1395, found 324.1377.

tert-Butyl (E/Z)-2-((4-((2,6-Dimethylphenyl)thio)but-3-en-1-yl) oxy)acetate (3da). Yellow oil, 84% yield (270 mg); ¹H NMR (400 MHz, CDCl₃): δ Z isomer 7.13–6.97 (m, 3H), 5.71 (dt, J = 9.3, 1.3 Hz, 1H), 5.64−5.55 (m, 1H), 3.93 (s, 2H), 3.57 (t, J = 6.8 Hz, 2H), 2.52 (qd, J = 6.8, 1.3 Hz, 2H), 2.40 (s, 6H), 1.42 (s, 9H); E isomer 7.13−6.97 (m, 3H), 5.87 (dt, J = 14.9, 1.3 Hz, 1H), 5.16−5.06 (m, 1H), 3.84 (s, 2H), 3.40 (t, $J = 7.0$ Hz, $2H$), 2.39 (s, $6H$), 2.27 (qd, $J = 7.0$, 1.3 Hz, $2H$), 1.40 $(s, 9H)$. ¹³C NMR (101 MHz, CDCl₃): $\delta Z + E$ 169.23, 169.17, 142.7, 142.1, 132.4, 130.2, 128.3, 128.0, 127.73, 127.69, 127.6, 127.3, 124.4, 124.3, 122.8, 81.1, 81.0, 70.5, 69.9, 68.3, 32.7, 28.8, 27.7, 27.6, 21.6, 21.2. HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₈H₂₆O₃S 322.1603, found 322.1619.

tert-Butyl 2-(4-oxo-3-((2-Fluorophenyl)thio)butoxy)acetate (4ga). Yellow oil, 59% yield (193 mg). 1 H NMR (400 MHz, CDCl₃): δ 9.45 (s, 1H), 7.37 (t, J = 7.0 Hz, 1H), 7.30−7.22 (m, 1H), 7.02 (t, J = 7.7 Hz, 2H), 3.88 (t, J = 10.4 Hz, 2H), 3.78 (t, J = 6.7 Hz, 1H), 3.71–3.59 (m, $2H$), 2.12 (dd, $J = 13.7, 6.1$ Hz, $1H$), 1.84 (dd, $J = 14.0, 5.6$ Hz, $1H$), 1.41 (s, 9H). ¹³C NMR (101 MHz, CDCl₃): δ 194.2, 169.5, 162.9 (d, J = 247.3 Hz), 136.7, 131.2 (d, J = 8.1 Hz), 124.7 (d, J = 7.7 Hz), 116.2 (d, J $= 23.2$ Hz), 81.7, 68.5 (d, J = 74.1 Hz), 53.5, 28.4, 28.1. HRMS (CI-TOF) m/z : $(M + H)^+$ Calcd for $C_{16}H_{22}O_4$ FS 329.1223, found 329.1217.

tert-Butyl 2-(4-Oxo-3-((4-fluorophenyl)thio)butoxy)acetate (4ha). Yellow oil, 91% yield (298 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.50 (d, $J = 2.8$ Hz, 1H), 7.47–7.37 (m, 2H), 7.03–6.96 (m, 2H), 3.93 (d, $J = 2.2$ Hz, 2H), 3.81−3.76 (m, 1H), 3.75−3.62 (m, 2H), 2.20−2.10 (m, 1H), 1.92−1.82 (m, 1H), 1.47 (s, 9H). 13C NMR (101 MHz, CDCl3): δ 194.1, 169.5, 163.2 (d, J = 249.5 Hz), 138.8 (d, J = 8.7 Hz), 136.5 (d, J = 8.4 Hz, 1H), 131.3 (d, J = 8.3 Hz), 116.4 (d, J = 22.0 Hz), 81.8, 68.8, 68.1, 54.3, 28.2, 28.1. HRMS (CI-TOF) m/z : M⁺ Calcd for C₁₆H₂₁O₄FS 328.1145, found 328.1154.

tert-Butyl 2-(4-Oxo-3-((2-chlorophenyl)thio)butoxy)acetate (4ia). Yellow oil, 61% yield (210 mg); $^1\text{H NMR}$ (400 MHz, CDCl₃): δ 9.51 $(dd, J = 2.8, 1.3 Hz, 1H), 7.44 (td, J = 7.7, 1.7 Hz, 1H), 7.36–7.29 (m,$ 1H), 7.13−7.05 (m, 2H), 3.93 (d, J = 4.6 Hz, 2H), 3.85 (td, J = 7.2, 2.8 Hz, 1H), 3.77−3.65 (m, 2H), 2.24−2.14 (m, 1H), 1.97−1.81 (m, 1H), 1.47 (s, 9H). ¹³C NMR (101 MHz, CDCl₃): δ 194.2, 169.4, 136.6, 131.2, 131.1, 124.7, 116.3, 116.0, 81.7, 68.8, 68.1, 53.4, 28.4, 28.1. HRMS (CI-TOF) $m/z\mathrm{:}~\mathrm{M^{+}}$ Calcd for $\mathrm{C_{16}H_{22}O_{4}CIS}$ 345.0927, found 345.0939.

tert-Butyl (Z/E)-2-((4-((3-chlorophenyl)thio)but-3-en-1-yl)oxy) *acetate (3ja).* Yellow oil, 24% yield (79 mg); δ 7.32 (dd, J = 6.4, 4.6 Hz, 1H), 7.27−7.16 (m, 3H), 6.30 (dd, J = 11.9, 10.5 Hz, 1H), 6.15−

6.00 (m, 1H), 4.02 (s, 2H), 3.66 (t, $J = 6.5$ Hz, 2H), 2.62 (ddd, $J = 13.6$, 6.8, 1.2 Hz, 1H), 2.58−2.52 (m, 1H), 1.52 (s, 9H). 13C NMR (101 MHz, CDCl3): δ Z + E 169.1, 169.0, 138.1, 137.8, 134.3, 134.2, 134.1, 130.6, 129.5, 129.4, 127.6, 127.3, 126.1, 125.8, 125.7, 125.6, 122.9, 121.6, 81.16, 81.10, 69.8, 69.7, 68.3, 68.2, 32.9, 29.1, 27.6. HRMS (CI-TOF) m/z: M⁺ Calcd for $C_{16}H_{21}O_3$ SCl 328.0900, found 328.0905.

tert-Butyl 2-(4-Oxo-3-((3-chlorophenyl)thio)butoxy)acetate (4ja). Yellow oil, 68% yield (234 mg); ¹H NMR (400 MHz, CDCl₃): δ 9.46 (d, J = 3.4 Hz, 1H), 7.55−7.34 (m, 2H), 7.28−7.14 (m, 2H), 3.93 (d, J = 3.8 Hz, 2H), 3.71 (t, $J = 5.7$ Hz, 2H), 2.30–2.20 (m, 1H), 2.11–1.93 (m, 1H), 1.47 (s, 9H). ¹³C NMR (75 MHz, CDCl₃): δ 193.4, 168.4, 135.5, 132.6, 129.1, 128.1, 126.4, 80.8, 67.8, 67.1, 52.0, 27.6, 27.1. HRMS (CI-TOF) m/z : $(M - C_4H_8)^+$ Calcd for $C_{12}H_{13}O_4$ SCl 288.0223, found 288.0219.

Hex-1-yne-1-d (2b'). ¹H NMR (400 MHz, CDCl₃): δ 2.19 (t, J = 6.9 Hz, 2H), 1.93 (s, 0.13H), 1.59−1.36 (m, 4H), 0.92 (t, J = 7.2 Hz, 3H).

2-(Phenylthio)hexanal-1-d (4**b**'). 1 H NMR (400 MHz, CDCl₃): δ 9.36 (d, J = 4.3 Hz, 0.13H), 7.42−7.34 (m, 2H), 7.33−7.27 (m, 3H), 3.51 (t, J = 7.2 Hz, 1H), 1.84−1.76 (m, 1H), 1.68−1.62 (m, 1H), 1.45− 1.28 (m, 4H), 0.92 (t, J = 7.1 Hz, 3H). HRMS (CI-TOF) m/z : M⁺ Calcd for $C_{12}H_{15}$ OSD 209.0985, found 209.0992.

2,2,6,6-Tetramethyl-1-((phenylthio)oxy)piperidine (**6**).¹² ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 7.75 - 7.63 \text{ (m, 2H)}, 7.50 - 7.34 \text{ (m, 3H)}, 1.84 \text{ (s,$ 1H), 1.67 (s, 4H), 1.57 (s, 4H), 1.51 (s, 3H), 1.48 (s, 3H), [1.3](#page-6-0)8 (s, 1H), 0.91 (s, 2H). MS (ESI-TOF) m/z : $(M + H)^+$ Calcd for C₁₅H₂₄NOS 266.2, found 266.2.

■ ASSOCIATED CONTENT

S Supporting Information

 1 H, 13 C NMR spectra for compounds 3 and 4. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b00641.

[■](http://pubs.acs.org) AUTHOR I[NFORMATION](http://pubs.acs.org/doi/abs/10.1021/acs.joc.5b00641)

Corresponding Authors

*E-mail: panxq@suda.edu.cn.

*E-mail: jpzou@suda.edu.cn.

Notes

The aut[hors declare no comp](mailto:jpzou@suda.edu.cn)eting financial interest.

■ ACKNOWLEDGMENTS

J.-P.Z. thanks the grant support by National Natural Science Foundation of China (Nos. 20772088, 21172163, 21472133) and the Priority Academic Program Development of Jiangsu Higher Education Institutions for financial support.

■ REFERENCES

(1) Selective recent publications: (a) Egami, H.; Sodeoka, M. Angew. Chem., Int. Ed. 2014, 53, 8294. (b) Chemler, S. R.; Bovino, M. T. ACS Catal. 2013, 3, 1076. (c) Muñiz, K.; Martínez, C. J. Org. Chem. 2013, 78, 2168. (d) Huang, S.-X.; Ding, K.-L. Angew. Chem., Int. Ed. 2011, 50, 7734. (e) Bataille, C. J. R.; Donohoe, T. J. Chem. Soc. Rev. 2011, 40, 114. (f) Minatti, A.; Muñiz, K. Chem. Soc. Rev. 2007, 36, 1142. (g) Chen, D. J.; Timmons, C.; Wei, H. X.; Li, G. G. J. Org. Chem. 2003, 68, 5742. (h) Chen, D. J.; Guo, L.; Liu, J. Y.; Kirtane, S.; Cannon, J. F.; Li, G. G. Org. Lett. 2005, 7, 921. (i) Li, G. G.; Kim, S. H.; Wei, H. X. Tetrahedron Lett. 2000, 41, 8699. (j) Liu, J. Y.; Wang, Y. N.; Li, G. G. Eur. J. Org. Chem. 2006, 3112. (k) Wei, H. X.; Siruta, S.; Li, G. G. Tetrahedron Lett. 2002, 43, 3809. (l) Zhi, S. J.; Han, J. L.; Chen, L.; An, G. H.; Pan, Y.; Li, G. G. Synthesis 2008, 1570. (m) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483.

(2) Selective recent publications: (a) Carboni, A.; Dagousset, G.; Magnier, E.; Masson, G. Chem. Commun. 2014, 50, 14197. (b) Xu, F.; Zhu, L.; Zhu, S.-B.; Yan, X.-M.; Xu, H.-C. Chem.--Eur. J. 2014, 20, 12740. (c) Dagousset, G.; Carboni, A.; Magnier, E.; Masson, G. Org. Lett. 2014, 16, 4340. (d) Kong, W.-Q.; Merino, E.; Nevado, C. Angew. Chem., Int. Ed. 2014, 53, 5078. (e) Oh, S. H.; Malpani, Y. R.; Ha, N.;

Jung, Y.-S.; Han, S. B. Org. Lett. 2014 , 16, 1310. (f) Peng, X.-X.; Deng, Y.-J.; Yang, X.-L.; Zhang, L.; Yu, W.; Han, B. Org. Lett. 2014, 16, 4650. (g) Lu, Q.-Q.; Liu, C.; Huang, Z.-Y.; Ma, Y.-Y.; Zhang, J.; Lei, A.-W. Chem. Commun. 2014 , 50, 14101. (h) Wang, F.; Wang, D.-H.; Mu, X.; Chen, P.-H.; Liu, G.-S. J. Am. Chem. Soc. 2014 , 136, 10202. (i) Deb, A.; Manna, S.; Modak, A.; Patra, T.; Maity, S.; Maiti, D. Angew. Chem., Int. Ed. 2013 , 52, 9747. (j) Lu, Q.-Q.; Zhang, J.; Wei, F.-L.; Qi, Y.; Wang, H.- M.; Liu, Z.-L.; Lei, A.-W. Angew. Chem., Int. Ed. 2013 , 52, 7156. (k) Su, Y.-J.; Sun, X.; Wu, G.-L.; Jiao, N. Angew. Chem., Int. Ed. 2013 , 52, 9808. (l) Zhang, H.-W.; Pu, W.-Y.; Xiong, T.; Li, Y.; Zhou, X.; Sun, K.; Liu, Q.; Zhang, Q. Angew. Chem., Int. Ed. 2013 , 52, 2529. (m) Zhang, C.-W.; Li, Z.-D.; Zhu, L.; Yu, L.-M.; Wang, Z.-T.; Li, C.-Z. J. Am. Chem. Soc. 2013 , 135, 14082. (n) Wang, Y.; Zhang, L.; Yang, Y.-H.; Zhang, P.; Du, Z.-T.; Wang, C.-Y. J. Am. Chem. Soc. 2013 , 135, 18048. (o) Zhou, M.-B.; Wang, C.-Y.; Song, R.-J.; Liu, Y.; Wei, W.-T.; Li, J.-H. Chem. Commun. 2013 , 49 , 10817. (p) Zhu, R.; Buchwald, S. L. Angew. Chem., Int. Ed. 2013, 52 , 12655. (q) Duan, X.-Y.; Yang, X.-L.; Fang, R.; Peng, X.-X.; Yu, W.; Han, B. J. Org. Chem. 2013, 78, 10692.

(3) (a) Dénès, D.; Pichowicz, M.; Povie, G.; Renaud, P. Chem. Rev. 2014 , 114, 2587. (b) Beckwith, A. L. J.; Wagner, R. D. J. Org. Chem. 1981 , 46, 3638. (c) Oswald, A. A. J. Org. Chem. 1961 , 26, 842. (d) Oswald, A. A.; Noel, F.; Fisk, G. J. Org. Chem. 1961 , 26, 3974. (e) Oswald, A. A.; Griesbaum, K.; Naegele, W. J. Am. Chem. Soc. 1964 , 86, 3791. (f) Szmant, H. H.; Nanjundiah, R. J. Org. Chem. 1978 , 43 , 1835. (g) Oswald, A. A.; Hudson, B. E., Jr.; Rodgers, G.; Noel, F. J. Org. Chem. 1962 , 27, 2439. (h) Oswald, A. A.; Greisbaum, K.; Hudson, B. E., Jr. J. Org. Chem. 1963 , 28, 2355. (i) Thaler, W. A.; Oswald, A. A.; Hudson, B. E., Jr. *J. Am. Chem. Soc.* **1965**, 87, 311. (j) Ueda, M.; Miyabe, H.; Shimizu, H.; Sugino, H.; Miyata, O.; Naito, T. Angew. Chem., Int. Ed. 2008 , 47, 5600. (k) Kim, J.; Li, H. B.; Rosenthal, A. S.; Sang, D.; Shapiro, T. A.; Bachi, M. D.; Posner, G. H. Tetrahedron 2006, 62, 4120. (l) O 'Neill, P. M.; Verissimo, E.; Ward, S. A.; Davies, J.; Korshin, E. E.; Araujo, N.; Pugh, M. D.; Cristiano, M. L. S.; Stocks, P. A.; Bachi, M. D. Bioorg. Med. Chem. Lett. 2006 , 16, 2991. (m) Amewu, R.; Gibbons, P.; Mukhtar, A.; Stachulski, A. V.; Ward, S. A.; Hall, C.; Rimmer, K.; Davies, J.; Vivas, L.; Bacsa, J.; Mercer, A. E.; Nixon, G.; Stocks, P. A.; O 'Neill, P. M. Org. Biomol. Chem. 2010, 8, 2068. ,

(4) (a) Ito, O. In S-Centered Radicals; Alfassi, Z. B., Ed.; Wiley: Chichester, 1999; p 193. (b) Griesbaum, K.; Oswald, A. A.; Hudson, B. E., Jr. J. Am. Chem. Soc. 1963 , 85, 1969. (c) Yoshida, J.; Nakatani, S.; Isoe, S. J. Org. Chem. 1993 , 58, 4855. (d) Ichinose, Y.; Wakamatsu, K.; Nozaki, K.; Birbaum, J. L.; Oshima, K.; Utimoto, K. Chem. Lett. 1987, 1647. (e) Beaufils, F.; Dénès, F.; Becattini, B.; Renaud, P.; Schenk, K. Adv. Synth. Catal. **2005**, 347, 1587. (f) Lachia, M.; Dénès, F.; Beaufils, F.; Renaud, P. *Org. Lett.* **2005**, 7, 4103. (g) Dénès, F.; Beaufils, F.; Renaud, , P. Org. Lett. 2007, 9, 4375. ,

(5) (a) Zhou, S.-F.; Li, D.-P.; Liu, K.; Zou, J.-P.; Asekun, O. T. J. Org. Chem. 2015 , 80, 1214. (b) Zhou, S.-F.; Pan, X.-P.; Zhou, Z.-H.; Shoberu, A.; Zou, J.-P. J. Org. Chem. **2015**, 80 (7), 3682.

(6) (a) Huang, C.-H.; Liao, K.-S.; De, S.-K.; Tsai, Y. M. Tetrahedron Lett. 2000 , 41, 3911. (b) Movassagh, B.; Yousefi, A. Monatsh. Chem. 2014 , 145, 1173.

(7) Aggarwal, V. K.; Eames, J.; Heras, M. A. D.; McIntyre, S.; Warren, S. J. Chem. Soc., Perkin Trans. 1 2000, 4456.

(8) Wang, W.; Li, H.; Wang, J.; Liao, L. X. Tetrahedron Lett. 2004 , 45 , 8229.

(9) Downey, C. W.; Craciun, S.; Southall, B. C.; Corsi, S.; Etchill, E. W.; Sault, R. J. Tetrahedron Lett. 2012, 53, 5763.

(10) Al-Masum, M.; Yamamoto, Y. J. Am. Chem. Soc. 1998 , 120, 3809. (11) Lenardão, E. J.; Silva, M. S.; Lara, R. G.; Marczewski, J. M.;

Sachini, M.; Jacob, R. G.; Alves, D.; Perin, G. ARKIVOC 2011 2, 272. , (12) Masafumi, U.; Hideto, M.; Hidenori, S.; Hisako, S.; Okiko, M.; Takeaki, N. Angew. Chem., Int. Ed. 2008 , 47, 5600.